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OIL CONTAMINATION IN OXYGEN SYSTEMS

Contract NObs-94416 Project No. SF013-08-14, Task 3917

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FOREWORD

This report covers the work performed under NAVSHIPS Contract No. NObs-94416 (Project No. SF013-08-14, Task 3917). The NAVSEC contract monitor was Mr. Herman Kraut.

The objective of this program was to collect the necessary experimental data on ignition and flame propagation in an cil contaminated oxygen system in order to more exactly define the required level of cleanliness for such systems. The results are to be incorporated into military specifications for oxygen systems to define the required cleanliness.

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INTRODUCTION

Under auspices of the Department of the Navy, NAVSEC, a study was undertaken to determine the compression ignition and flame propagation limits of hydrocarbons in a high pressure oxygen system.

Considerable work has been done concerning compression ignition and flame propagation in high pressure air systems, but very little has been done with pressurized oxygen systems. The limited work performed with oxygen did not employ conditions corresponding to the specific systems now in use, namely 3000 psi oxygen with 1/2" monel piping. The program reported herein was designed to study that specific area

The study was primarily concerned with the determination of the acceptable levels of oil contamination (2190) on the internal surface of a 1/2" monel pipe below which compression ignition and flaw propagation will not occur when pressurized with 3000 psi oxygen. It was also desired to determine that level of oil contamination where, upon ignition, a pressure rise will not exceed 25%.

The first phase of the program consisted of a literature survey to assemble and review all available information from previous studien in the area of compression ignition and flame propagation in high pressure air and oxygen systems. A summary of these earlier studien is presented in Section I.

Following the literature survey a test program was designed and initiated to accomplish the objectives of the contract. A description of the test apparatus used is presented in Section II, the testing procedure is described in Section III, and the results of the test program are presented in Section IV.

Throughout this report, the term "oil concentration" is used. This refers to the amount of oil coating, expressed as mg/ft², on the inside surface of the test pipe.

SUMMARY

A test program was conducted to determine the ignition limits of eil contamination in 1/2" monel pipe when pressurized with 3000 pri oxygen. The oils investigated were 2190 lubricating oil and Habcool #318 eutting oil. Thin films of these oils were applied to the internal surface of a 4 ft. section of 1/2" monel pipe. The pipe was then rapidly pressurized with 3000 psi oxygen. In addition, tests were made using oil droplets and varying the initial pipe wall temperature.

The test results showed little danger of compression ignitions at oil concentrations below 160 mg./ft² of 2190, or below 110 mg./ft² fer Habcool. Concentrations of both oils above 260 mg./ft² will ignite when rapidly pressurized with 3000 psi oxygen. These concentrations must be compared to the present oxygen clean standards of 0.013 mg./ft² maximum allowable hydrocarbon concentration.

The results also showed that ignition will not occur with oil drople's or puddles smaller than 100 pt in size and that a pressure rise greater than 25% will not occur below oil concentrations of 500 mg./ft2.

As a result of this test program, it is recommended that the oxygen clean standards be relaxed to 1.5 mg./ft². The level of oil concentration is a factor of 100 below the lower ignition limits defined by this test program. It is felt that such a safety factor more than compensates for any variation in the lower ignition himly due to effective pipe length, wall temperature, and non-uniformity is oil contamination.

I LITERATURE SURVEY

In recent years the phenomenon of compression ignition in pneumatic systems has been extensively studied. Such ignitions are induced in systems contaminated with sufficient quantities of a combustible organic material and exposed to rapid pressurization with an oxidizing gas.

Most of the work was concerned with high pressure air systems contaminated with lubricating oil from air compressors. Faeth and White conducted a series of tests in which they investigated the combustical characteristics of 25 mg. samples of various lubricating oils placed at the dead end of a pipe emposed to rapid pressurization with air. They defined the combustion ranges as functions of various parameters including the rate of pressurization, air pressure, length of pipe, and the initial ambient system temperature. The results of these tests quantify—the expected trends due to variation in the parameters. Of interest was the fact that the occurrence of combustion was greater, the shorter the length of pipe used down to a length of 1 ft. However, the change in combustion limit was only a mild function of the pipe length.

Wilson (2) and co-workers investigated the compression ignition characteristics of lubricating cils evenly coated on the internal surface of a 1 1/2" diameter test pipe. Using a 15 ft. length of pipe, they studied both auto-ignition and flame propagation using 2190 and Callulute lubricating cils. Wilson defined the lower combustible limit in the range of 1000 mg./ft² of oil. He also found that used degraded oil was much more reactive than new unused oils - attributing this to breakdown products in the oil. This conclusion was later verified by Eabstakis (3). Both Wilson (2) and Perice (4), concerned with this type of ignition caused by degraded oil blow-by from air compressors, recommend proper cleaning and maintenance to prevent dangerous conditions from arising.

Facth⁽¹⁾ did a theoretical analysis of the rapid compression processin an attempt to predict whether conditions for ignition would exist in a given system. Predictions based on his analysis compared well with his observed experimental data.

Dallinger and McGill⁽⁶⁾, concerned with high pressure air systems aboard submarines, surveyed work done in this field. They reported that an increase in pipe diameter increased the temperatures obtained during a rapid compression, thus lowering the concentration of oil necessary for ignition.

Russian interest in the problem of compression ignition with oxygen is indicated by the work of Gordeyev and coworkers (7). They reported lower explosive limits of 26,000 mg./ft² of oil and lower combustion limits of 190 mg./ft². However, they used oils with Russian designations, making comparison with 2190 difficult, and they falled to report pertinent parameters such as rates of pressurization and pipe diameters.

The dangers inherent in high pressure air systems, due to compressionignition, lead to questions about parallel phenomenon in high pressure oxygen systems. Baum and his coworkers investigated 7500 psi oxygen systems, looking at effects of system temperature, vibration, shock, extended storage, contamination, and material compatibility. In regard to ignition, Baum are recommended absolute cleanliness from organics, and homogeneous hydrocarbon concentrations less that 50 ppm.

In 1961, Kehat ⁽⁹⁾ studied ignition and flame propagation at pressure; up to 1500 psi using gaseous oxygen. Using a 1/4" test pipe event; coated with oil, he used a spark and an electric match to induce ignition. Kehat found no significant danger of ignition at concentrations up to 1000 mg./ft² of oil. Also of interest was Kehat i technique of evenly coating a small diameter pipe, 1/2", with oil namely of dissolving the oil in a solvent, carbon tetrachloride, applying the solution to the pipe, and then allowing the solvent to evaporate, leaving a thin film of oil behind.

A summary of the literature survey is presented in Section 4.2.2 and compared with the recommendations of this current test program.

II TEST APPARATUS

2.1 General Description

The basic test apparatus is illustrated schematically in Figure 1.

The test apparatus basically consisted of a four foot section of 1/2 inch, schedule 80, monel pipe plugged at one end. The high pressure caygen was supplied from a two tank system. One tank (15 ft³ capacity) at 3800 psig was used to supply the second tank (9 ft³) at 3000 psig which, in turn, supplied the test pipe. The instrumentation and control equipment consisted of flame probes, pressure transducers, thermocouples, external strain gauges, a 5000 psi relief valve, a gas sampling bomb, a manually operated ball valve, a quick opening Marotto valve, a check valve and a manually operated vent valve. Photographs of the test pipe, in place for a run, are shown in Figures 2, 3, 4, and 5. A more detailed description of the components is found in Table I.

A four foot length of pipe was selected as the optimum length. This selection is a compromise between the minimum length recommended by Faeth and White and the physical instrumentation requirements necessary to measure flame propagation.

This test apparatus was modified slightly during the test program to accommodate a sparking device. This alteration consisted of removing the pressure transducer (P_1) closest to the dead end and replacing it with the sparking device. This modification is shown in Figure 6.

In addition to the pipe just described, another test pipe was used during the final portion of testing with a minimum of instrumentation located adjacent to the dead end. This second pipe was also a 4 ft. length of 1/2" monel pipe, but it had only two instrumentation taps and a vent line. This test set up is shown schematically in Figure 7. Photographs of this test set up are shown in Figures 8 and 9. As can be seen in the figures a nichrome heating coil was added to the pipe to give the capability of operating at elevated initial temperatures.

TABLE I LIST OF APPARATUS AND COMPONENTS

	Component	<u>Description</u>
1	3000 psig 02 supply flask	9 ft ³ , MTL-C-1511A
1	3800 paig O2 supply flask	15 ft ³ , MIL-C-2809B
1	Solenoid valve	Marotta M.V. 173
1	Pressure gauge	Q to 10,000 psig scale
1	Check valve	3/4"
3	Nipples	3/4" pipe, NiCu
1	Pipe union	1/2", NiCu
1	Pipe cap	1/2", NiCu
1	Flame arrestors	1/2" I.P.S. NiCu
1	Pipe tee	3/4" monel
1	Test pipe	4 ft. Sch. 80, 1/2" Monel MIL-T-1368 Ty. 1, Cond. 1
1	Relief valve	5,000 psi relieve pressure
1	Ball valve	3/4", manual
1	Vent valve	1/1/", manual
1	Gas sampler bomb	Sec Section 2.2
Ì,	Strain gauges	n
3	Pressure transducers	11
3	Temperature probes	!!
•	-	11

The flame arrestor used in Runs A-1 through A-24, Runs D-1 through D-5, and F-36 was made of stainless steel.

Flame probes

For safety considerations, the test pipe was located in a concrete test pit and was surrounded by sand bags. The control and instrumentation read-outs were located in an adjacent test pit.

2.2 <u>Instrumentation</u>

A STATE

Instrumentation on the basic test set-up consisted of four flame probes spaced 12 inches apart; three pressure transducers, 18 inches apart; four external strain gauges; and three thermocouples, 18 inches apart. The locations of this instrumentation can be seen in the figures presented in the previous section.

In addition to this basic instrumentation, a sparking device was used in an attempt to induce ignitions. The location of this spanking device can also be seen in the figures presented in the previous section. The outputs of the instrumentation probes were recorded on an oscillograph recorder with a paper speed of 64 inches/sec.

A close-up of a flame probe, a thermocouple, and the sparking device is shown in Figure 10; while a photograph of a pressure transduce; a strain gauge installation and the flame arrestor is shown in Figure 11.

A block diagram of the basic instrumentation is presented in Figure 12, while a schematic of the control circuitry is shown in Figure 13. A detailed list and description of the major instrumentation components is presented in Table II. The time sequence in which all the instrumentation and control circuitry operated is shown in Figure 14.

The instrumentation block diagram (Figure 12) shows the relationship of the control circuitry and the instrumentation probes to the recording device. The flame probes, the pressure transducers, and the strain gauges are energized by the circuitry; signals detected by them are then transmitted to the recorder. The thermocouples, using an ice water bath as a reference, generate voltages indicative of their temperatures and transmit them to the recorder. The spacking

TABLE II LIST OF INSTRUMENTATION EQUIPMENT

	Description	Manufacturer
1.	Chromel/alumel thermocouples (400 milliseconds response time)	Thermoelectric, Inc.
2.	Chromel wire flame probes - porcelain base with 300 volt operating potential (< 1.7 millisecond response time)	Electric Boat Division (not commercial)
3.	Pressure transducers - 10,000 lbs/in ² , air cooled type (.05 millisecond response time)	Norwood, Model 102
4.	UV light beam oscillograph recorder - galvonometer (1.7 millisecond response time)	Consolidated Electrodynamics Corp. (C.E.C.), Model #124
5.	3 kc. carrier amplifiers for recorder (15 units)	C.E.C., Model #1-113B
6.	Modulators for amplifiers (7 units) (2 milliseconds response time)	C.B.C., Model #15-605
7.	Oscillator power supplies (2 units)	C.E.C., Model #2-105B
8.	Variable AC Voltage Supply	Superior Electric
9.	Displacement Transducer (< 1.7 millisecond response time)	Underwater Explosion Research Division (USN)
10.	Potentiometer and Switch (Auxilliary temperature measurements)	Leeds & Northrup
11.	120 OHM Strain Gauges	Baldwin-Lima-Hamilton

device discharges the electrical energy stored in its capacitoms (Figure 13B) and this event is transmitted to the recorder. The potenticmeter attached to the stem of the Marotta valve is energiaed by the circuitry and transmits the change in position of the valve to the recorder. The Marotta valve itself is activated by the control circuitry; while time delay relay 1 (TD-1) and (TD-2) are activated by the same circuitry and in turn control parts of this circuitry (Figure 13).

As seen in Figure 13A, the circuitry is activated by closing switch S-1 which energizes the indicating light plus the flasher and bell to warn personnel in the area that a test is in progress. A test is initiated by closing S₃ which activates relay K₁ and TD-1. Activation of relay K₁ closes contacts K₁₋₁ which keeps itself activated without regard to the position of S₃. Activation of TD-1 closes the contacts TD1-1 which, in turn, activates TD-2, TD-3, and opens the Marotta valve. Activation of TD-2, after a preset period of time, shuts down the circuitry. Activation of TD-3, after a preset period of time, opens contacts TD3-1, closing the Marotta valve.

The circuitry of Figure 13B is activated by the closing of TD3-2, which has been closed by TD-3 of Figure 13A. With the contact TD3-2 closed, TD-5 is activated which in turn closes contacts TD5-1 activating K_2 . K_2 closes contact K_{2-1} which discharges the spark. The 600 ρ capacitors are charged by closing switch S_4 . Contacts TD3-3 isolate this circuit while TD-3 is activated.

III TEST PROCEDURE

3.1 Preliminary Tests

Before proceeding with ignition tests, it was necessary to verify the proposed oil coating technique.

Preliminary tests were conducted to determine whether the oil coalling technique selected would be satisfactory. This technique consisted of dissolving the oil in a low-boiling, non-combustible solvent Freon TF and then applying the resultant solution to the pipe. Upon evaporation of the Freon, a residual coating of oil was left on the pipe. In evaluating this technique, a sacrificial section of pipe was used. After the coating operation, the pipe was cut into small sections and the amount of oil present on each section was determined by chemical analysis. This analysis consisted of removing the oil from the pipe segments with a low boiling point solvent, then evaporating off the solvent. The amount of oil removed from each segment was then determined gravimetrically.

This same method of evaluation was used to determine the feasibil' of other oil coating techniques as part of the verification. The other oil coating techniques evaluated were: a) applying the oil directly with a circular brush or swab, b) spraying the oil onto the pipe, c) partial removal of a heavy oil coating using a solvent.

3.2 Test Plan

The test plan included the following areas of investigation:

1) The costing of the test pipe with various concentrations of 2190 oil at ambient temperatures and attumpting a compression ignition with 3000 psi oxygen; the oil coatings were increased in small increments in concentrations ranging from 0.08 mg/ft² to 640 mg/ft².

2) Placing droplets of 2190 oil near the dead end of the test pipe

and attempting a compression ignition with 3000 psi oxygen; the size of the oil droplets used ranged from 5 microliters (46) to 90.4%. 3) Employing the combination of coating the test pipe with 2190 oil and placing a droplet of the oil near the dead end of the pipe, and attempting a compression ignition with 3000 :.. coygen; oil coating concentrations ranged from 12 mg/ft2 to 88 mg /m2 while the oil droplets ranged in size from 10 gl to 100 gl . It) contains the test pipe with various concentrations of 2190 oil or of Habeet cutting oil, using an electrical spark to induce ignition in a 3000 psi oxygen atmosphere; the 2190 oil concentrations reaged Tax 0.16 mg/ft2 to 640 mg/ft2, while the Habcool concentrations ranged from 100 mg/ft2 to 480 mg/ft2. 5) Coating a modified test pipe (minimum instrumentation) with various concentrations of 2190 old and of Habcool cutting oil, and attempting a compression ignition with 3000 psi oxygen; the 2190 oil concentrations ranged from 30 mg/jt? to 480 mg/ft2, while the Habcool concentration ranged from 160 mg/ot to 480 mg/ft².

The 2190 (MIL-L-17331A) lubricating oil was selected to be used during this test program because most of the previour investigations of compression ignition and flame propagation were performed using this lubricating oil. Thus, in order to compare results with the earlier work, 2190 was used in this investigation. Habcool #318 cutting oil was selected as a typical organic cutting oil used in a machine shop with which piping might be contaminated during fabrication in actual practice.

3.3 Testing Procedure

In making a test run the following procedure was used:

- 1. The test pipe was thoroughly cleaned by use of Freon (TF) ringer and a lint-free cloth.
- 2. With the instrumentation probes replaced by plugs, the pipe war filled with an oil/Freen solution of a known concentration.
- 3. The oil/Freon solution was emptied and the remaining Frech allowed to evaporate (15 to 20 minutes was allowed) leaving a thin, uniform oil film.
- 4. The instrumentation probes were replaced into their proper positions.
- 5. The test pipe was placed "on-line", putting the flame arrestor into position within the union.
- 6. The calibration of the instrumentation was checked as well as the electrical control circuitry.
- 7. The position of all hand-operated valves was checked.
- 8. The test area was cleared of all personnel and the test was made.
- 9. After the run was completed, the electrical power was turned off and all valves secured.
- 10. A gas sample was taken for analysis and the test pipe was vented down to ambient pressure.
- 11. The test pipe was disconnected and was visually inspected.
- 12. The test pipe and instrumentation probes were thoroughly cleaned in preparation for the next run.

The gas sample at the conclusion of a test was analyzed for carbon dioxide, hydrogen, carbon monoxide, exygen and nitrogen waing gas chromatography.

No attempt was made to replace the air in the test pipe with pure 0, before the rapid pressurization. This initial air is compressed to within 1/4" of the dead end during compression from 1%.7 psi to 3000 psi and it was felt that this small amount of nitrogen, in comparison to the amount of oxygen present, would have a negligible effect on the results.

IV TEST RESULTS

4.1 Preliminary Tests

The oil coating tests showed that the selected technique of dissolving the oil in a non-combustible solvent, Freon TF, gave the best results. The oil was evenly coated onto the surface of the pipe and the concentration of oil on the pipe was proportional to the concentration of oil in the Freen. This relationship is shown in Table III. It was also found to be the best technique from an operational point of view - being less prone to technician error and having good reproducibility.

TABLE III
OIL COATING CONCENTRATIONS
(1/2" Monel Pipe)

Oil Concentration in Freon - PPM	Oil Coating on Pipe - mg/ft ²	Oil Film Thickness - mincens
5	0.0125	1.2×10^{-3}
10	0.025	2.4 x 1.0 ⁻⁵
100	0.25	2.4.x 10 ⁻²
1000	2.5	0.24
10,000	24.6	2.4
50,000	123.0	12.0
100,000	246.0	24.0
200,000	492.0	48.0
3 70,000	730.0	72.0

4.2 Ignition Test Results

The significant data are presented in Section 4.2.1 and Appendix A, while an interpretation and discussion of those data is found in Section 4.2.2.

4.2.1 Data

The parameters measured during each of the test runs are presented to Appendix A. The parameters include: oil concentration, smillent temperature, maximum pressure, rate of pressure rise, include temperature flame probe saturation, and strain gauge indication. The temperature is presented are the maximum temperatures recorded. Due to the bid of and response time of the thermocouples, the actual temperatures enjoying and are much higher.

It should be noted that at various points throughout the test programals blank runs were made - i.e. runs in which no oil was put into the test pipe. These runs were made to determine the time historian of the the various parameters due only to the rapid pressurization of the test pipe with oxygen, and to serve as a check on the instrumentation. The results of all the blank runs were identical, therefore, the results of only one of these runs is presented.

4.2.2 Correlation and Interpretation of Results
The first test run with oil contamination resulted in a violent explosion causing extensive damage to the test apparatus. This damage is shown in Figures 15-20. A comparison can be made with the undamaged test pipe shown in previous sections.

As can be seen in the figures a two inch section of the test pipe, at the inlet end, was disintegrated along with part of the coupling; one of the pressure probes was blown out of its fitting; several of the electrical connectors to the instrumentation probes were burned; and the copper "O" ring seals on the two remaining pressure transducers were blown out. In addition, the flame arrestor was distintegrated and the check valve welded closed. It should be noted that for this run the coupling at the inlet end of the test pipe and the flame arrestor were both made of stainless steel.

After the explosion, it was felt that the level of oil contamination. 160 mg/ft², was well into the explosive range and that the testing should be conducted in a much lower range of oil concentration. Subsequent test runs showed, however, that levels of oil contamination higher than 160 mg/ft² can withstand rapid pressurization by pure oxygen without resulting in an explosion. These subsequent tests employed monel and nickel-copper components only, while the initial run at 160 mg/ft², which resulted in an explosion, had a stainless

steel flame arrestor. In an attempt to duplicate the explosive ring a stainless steel flame arrestor was put into the system (Run F-30). In this run, though the test pipe did not rupture, there was an ignition and flame propagation, as opposed to no indications of ignition or flame propagation under the same conditions using a month flame arrestor (Run F-18).

No completely satisfactory explanation of the explosion is available, but it is significant that no indications of ignition were obtained with the monel flame arrestor, while very definite indications were observed with the stainless steel flame arrestor.

As a result of the explosion in the first run, subsequent tests were conducted (Runs A2-A24, F1-F28, F35) by evenly coating the test pipe with 2190 oil with concentrations of 0.08 mg./ft² up to 160 mg./ft² without any indication of ignition. At oil concentrations between 173 mg/ft² and 266 mg./ft² there were indications of ignition and flame propagation on some instrumentation probes but not on others. At oil concentrations above 266 mg./ft², there were definite indications of ignition and flame propagation. These results can be seen graphically in Figure 21.

On the supposition that it was possible to have sufficient oil in the test pipe to propagate a flame while not having a high enough concentration to cause an auto-ignition, a series of runs were conducted in which it was attempted to induce an ignition. Thus runs (Runs Dl-Dl4, ADl-ADlO) were made with small drops of oil placed near the dead end of the test pipe (thus giving a high local concentration of oil). Oil droplets ranging in size from 5 l to 100 l were used alone and incombination with a uniform oil film ranging in concentrations from 12 mg./ft² to 88 mg./ft². In Run D-12, the gas analysis after the run revealed the presence of CO2, indicating combustion of the oil. However, the other instrumentation did not confirm this combustion, and the results could not be duplicated in subsequent tests at similar or more severe conditions (Runs AD-8 and AD-10).

In the test runs with oil droplets of 70_{M} or greater, H_2 was do a in the gas after the test run. It is believed that this H_2 is a product of the thermal decomposition of the oil. However, oil are lets were used only in an attempt to induce ignitions. This they failed to do, so the appearance of H_2 , though interesting, was not investigated further. However, two things should be noted in this regard: 1) the occurrence of oil droplets, 70_{M} or greater, in a clean oxygen system is highly unlikely and 2) the H_2 appeared in concentrations well below its lower ignition limit.

Having failed to induce ignition using small droplets of oil, an attempt was made to induce ignition by means of an electrical spant. A series of runs were made using the sparking device (Runs F1-F28, F35, F36). In those runs in which ignitions were observed, they were of the compression ignition type, and involved oil concentrations above 160 mg./ft². The spark did not initiate any ignitions nor affect flame propagation in any way.

Test runs made in which the pipe was evenly coated with Habcool at 70°F (Runs F29-F34) indicated that the no ignition range was below about 110 mg./ft², while the ignition range was somewhere above 250 mg./ft². Comparison of these results with those obtained for 2190 (Figure 21) show Habcool to be somewhat more reactive.

In an attempt to attain higher temperatures from the rapid compression with pure oxygen, a series of tests were run using a pipe with a minimum of instrumentation located adjacent to the dead end. Because of the minimum amount of instrumentation, it was more difficult to get accurate indications of ignition and flame propagation. However, from the instrumentation available, the following results were obtained:

A. Using 2190 oil at 70° F, tests were run at concentrations of 80 mg./ft² to 480 mg./ft² (Runs Cl-C6). No ignition was obtained

below 160 mg/ft², while the ignition range was found to be now.

200 mg/ft². These results approximate the results shown in

Figure 21.

- B. Using 2190 oil heated to 125°F, tests were run at concentrations ranging from 160 mg/ft² to 480 mg/ft² (Runs 07-011). Indications of ignition were seen at the lower concentration of 160 mg/ft, indicating that the level of oil contamination necessary for ignition and flame propagation are somewhat lower at 125°F this at 70°F.
- c. Using Habcool cutting oil at 70°F, tests were run at concentrations ranging from 160 mg/ft² to 480 mg/ft² (Runs Cl2-Cl6). Definite ignitions were obtained above 200 mg/ft², while the non-ignition range is somewhere below 160 mg/ft². The non-ignition and ignition ranges are about the same as those found with the regular test pipe.
- D. Using Habcool oil at 125°F, tests were run at concentrations ranging from 160 mg/ft² to 480 mg/ft² (Nuns C17-C21). The results from these tests were not distinguishable from those at 70°F.

A more quantitative correlation can be made by plotting the resultant % CO₂, the maximum recorded temperature, and maximum pressure obtained as functions of level of oil concentration. Such plots are shown in Figures 22, 23 and 24. It can be seen that these results match the non-ignition, transient, and ignition ranges earlier defined more qualitatively.

The carbon dioxide found was considerably less than would be dictated with 100% combustion of the oil. This is, however, consistent with the fact that carbon deposits and residual oil were found in the beau pipe after ignitions, indicating considerably less than complete combustion.

It is possible to compare these results with those obtained by earlier investigators. Kehat reports little danger of ignition in oxygen

systems up to 1500 psig at concentrations up to 1000 mg/ft². The compares to our lower ignition limit of 160 mg/ft² at 3000 psi.

Kehat used a pure compound, n-hexadecine, for his work, while a work was performed with a commercial lubricating oil, 2100.

Comparison of the properties of n-hexadecine and 2190 - e.g., molecular weights, auto-ignition temperature - and considering the differences in the oxygen pressures used reveal our results to in reasonable agreement with Kehat.

It is also possible to compare our results with the results of Wilson et al. Wilson reports the lower ignition limit in high pressure air systems at 3000 psig of 2190 oil to be 1000 mg/ft². The extrapolating these results to pure oxygen at 3000 psig, a lower ignition limit would be approximately 200 mg./ft². This is in good agreement with the current results.

A summary comparison of the current results and those obtained by earlier investigators is presented in the following chart:

Investigator	Test Gas	Test Apparatus	Oll Used	Lower Ignition bimit
Kehat	0 ₂ , 1500 psi	1/2" pipe	crs _H se	1.000 mp/3 t ²
Wilson	Air, 3000 psi	1-1/2" pipe	2190	1000 m3/31/2
Gordeyev	0 2	-	-	190 mg/ft ²
Presti	0 ₂ , 3000 psi	1/2" pipe	21.90	160 mg/ft ²
Baum	o ₂ , 1500 psi	(50 ppm homogened recommended lower	ous mixture or limit)	-

V CONCLUSIONS AND RECOMMENDATIONS

From the test results obtained the following conclusions can be drawn for 3000 psi oxygen systems:

- 1. Compression ignition and flame propagation will not ceeder below oil concentrations of 160 mg/ft² at oil temperatures below 125°F in a 1/2", schedule 80, monel pipe.
- 2. Compression ignitions will not occur with oil droplets or puddles smaller than 100 Alin size.
- 3. Haboool cutting oil was found to be somewhat more reactive than 2190.
- 4. Present oxygen clean standards (as specified by Shipyard Standard Practice 1.9 Rev. F, Oxygen and Nitrogen Systems, paragraph 3.2.3.1d) which require oil film concentrations of 0.613 mg/ft² or less, are much too rigorous and can be relaxed. The oil film concentration of 0.013 mg/ft² corresponds to obtaining a concentration of 5 ppm in a Freon rinse, while 160 mg/ft² corresponds to 65,000 ppm in Freen.

It is recommended that the allowable oil film concentration specified by the above Shipyard Standard Practice be set at 500 ppm oil concentration in a freen rinse. This concentration allows a cafety factor of 100 over the values at which oil ignitions begin to occur.

It is felt that such a safety factor more than adequately allows for any variation in the lower ignition limit due to variations in effective pipe lengths below 4 feet which might be found in an actual pressurized oxygen system.

5. To obtain pressure rises greater than 25% requires oil concentrations greater than 500 mg./ft. This can be seen from Figure 2%.

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APPENDICES

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			ambi-		MAX		PRESI			x Tim	Š		RATE OF PRESSURE RISE INSIDE PIPE	MAXIMIS METER (1)		
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/9	C=A	320	70	2960 in 104	-	-	74000		
5/9	C≈5	400	70	3050 in 118			hocco		
j/9	C6	480	70	3240 in 96	11.4		45000	· !	<i>f</i>
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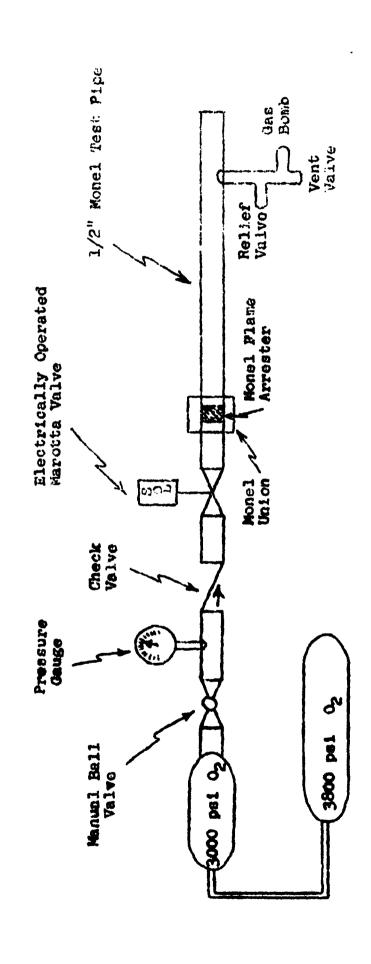
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DIAGRAM OF TEST APPARATUS

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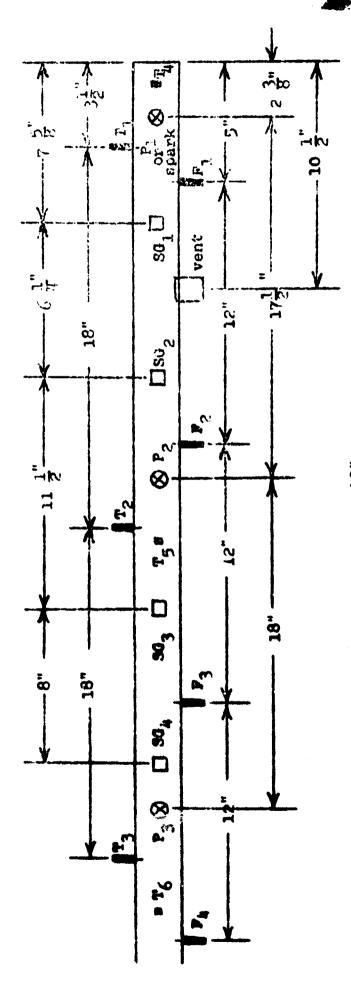
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T - Thersocouple

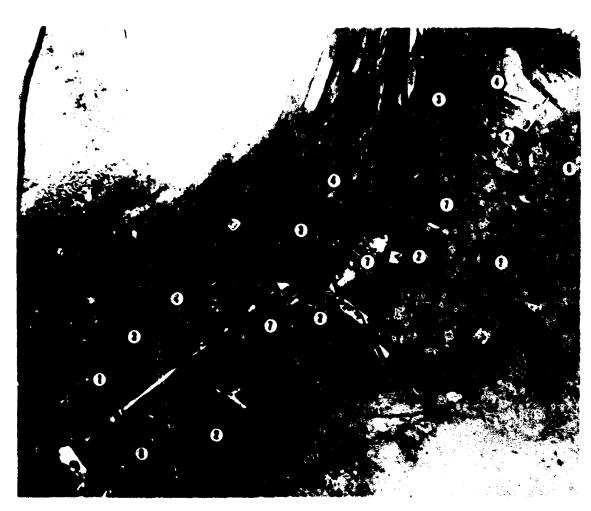
P - Plame Probe

P - Pressure Transducer

SG - Strain Gauge



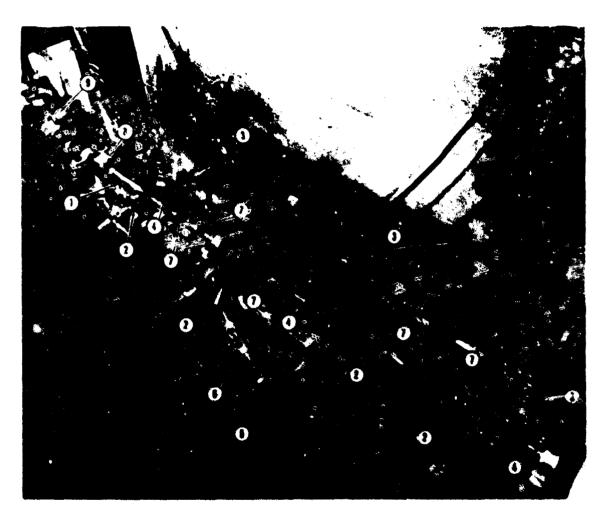
Pipe Length: 48"



TEST PIPE

- 1) 1/2" MONEL PIPE
- 2) FLAME PROBES
- 3) TEMPERATURE PROBES
- 4) PRESSURE TRANSDUCERS

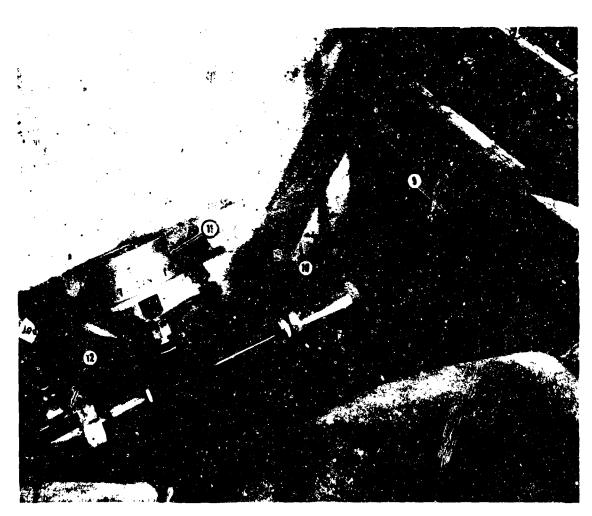
- 5) RELIEF VALVE
- 6) GAS SAMPLE BOMB
- 7) STRAIN GAUGES
- 8) MONEL UNION (FLAME ARRESTER INSIDE)



TEST_PIPE

- 1) 1 2" MONEL PIPE
- 2) FLAME PROBES
- 3) TEMPERATURE PROBES
- 4) PRESSURE TRANSDUCERS

- 5) RELIEF VALVE
- 6) GAS SAMPLE BOMB
- 7) ETRAIN GAUGES
- B) ME HEL UNION
- 9) MAROTTA VALVE



OXYGEN SUPPLY PIPING FIGURE 4

- 9) MAROTTA VALVE
- 10) CHECK VALVE
- 11) PRESSURE GAUGE
- 12) HAND OPERATED BALL VALVE



OXYGEN SUPPLY PIPING FIGURE 5

- 10) CHECK VALVE
- 11) PRESSURE GAUGE
- 12) HAND OPERATED BALL VALVE
- 13) 3000 PSI OXYGEN SUPPLY TANK



TEST PIPE WITH SPARKING DEVICE FIGURE 6

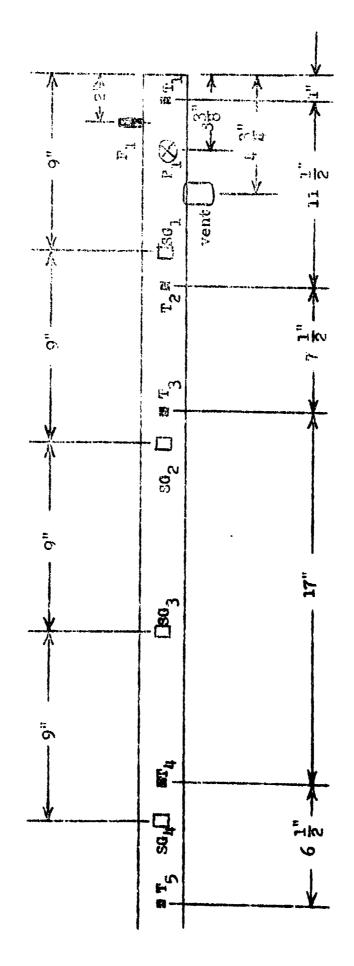
- 1) SPARKING DEVICE
- 2) FLAME PROBE
- 3) TEMPERATURE PROBE
- 4) STRAIN GAUGE
- 5) VENT LINE

T - Thermonouple

B - Flane Probe

P - Prussure Transducer

SG - Strein Gauge



Pipe Length: 45 1"

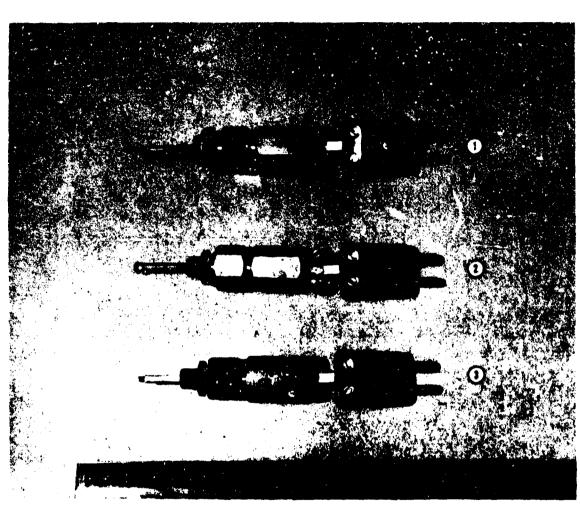


MODIFIED TEST PIPE FIGURE 8

- 1) PRESSURE TRANSDUCER
- 2) FLAME PROBE
- 3) RELIEF VALVE 4) GAS SAMPLING BOMB
- 5) STRAIN GAUGES 6) HEATING WIRE
 - 7) MONEL UNION

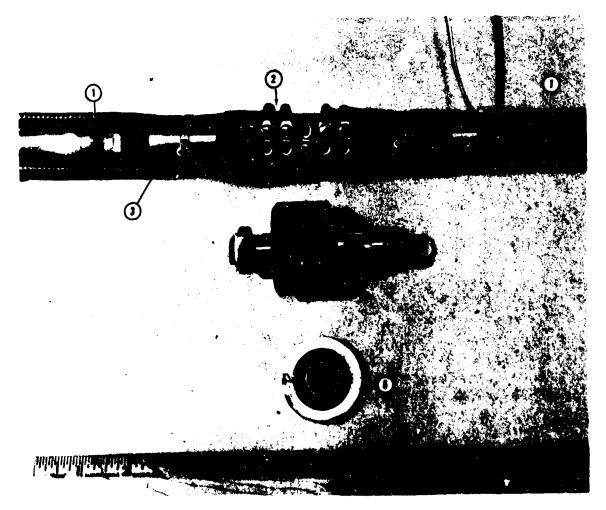
MODIFIED TEST PIPE FIGURE 9

- 1) PRESSURE TRANSDŲCER
 - 2) FLAME PROBE
- 3) RELIEF VALVE
 4) GAS SAMPLING BOMB
 - 5) STRAIN GAUGES
- 6) HEATING WIRE
- MONEL UNION
 MAROTTA VALVE



INSTRUMENTATION PROBES FIGURE 10

- 1) SPARKING DEVICE
- 2) THERMOCOUPLE
- 3) FLAME PROBE



TEST PIPE COMPONENTS
FIGURE 11

- 1) STRAIN GAUGE
- 2) TERMINAL FOR STRAIN GAUGES
- 3) HEATING COIL
- 4) PRESSURE TRANSDUCER
- 5) FLAME ARRESTOR

Cenérator Modulator Csoillograph Transaucer Lisplacement Attenestor Calibration Amplifier Concrol Board Master Abbreviations: Att. -Ogl. -Olepl. och. -Mr. nn. 1 5000 . Sec. ·am, INSTRUMENTATION BLOCK DIAGRAM - FIGURE 12 -Channel Ultra-Record-C.B.C. violet Osc. OIL CONTAMENATION IN OXYGEN SYSTEMS C.E.C. 3KC Carrier System Amp. C.B.C. Spark Gen. & Att. Wetwork Superior formonia Att. Hetwork Notwork Cal. & Network Network Col. & Percr Att. Cal. Walve Valve conples Preseure Thermo-Strain Device Trans. Displ. Trans. Spark Oageo

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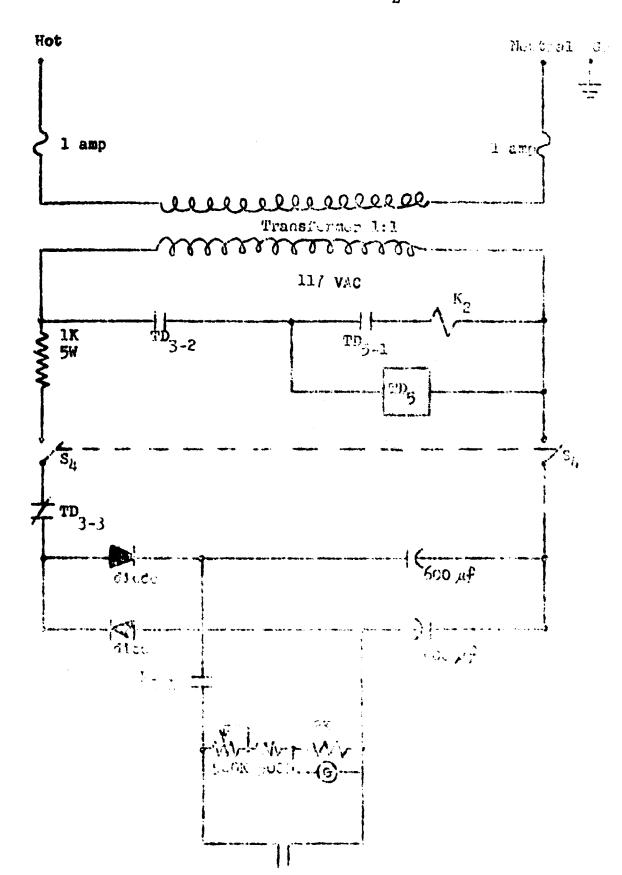
OII. CONTAMINATION IN OXYGEN SYSTEMS AUTOMATIC REMOTE CONTROL CURCULT

FIGURE 13A Het Routral Johns 117 VAC 60 H₂ geeleeleeleeleege Transformer 1:1 -CORRELEGICO BARRAGO 117 VAC 5 amp 5 amp Sı light bulb lasher bell TD₂₋₁ Caretta 7°C 1

light bilb

to recorder

OIL CONTAMINATION IN OXYGEN SYSTEMS SPARK GENERATION CIRCUIT FIGURE 13B 117 VAC 60 HZ



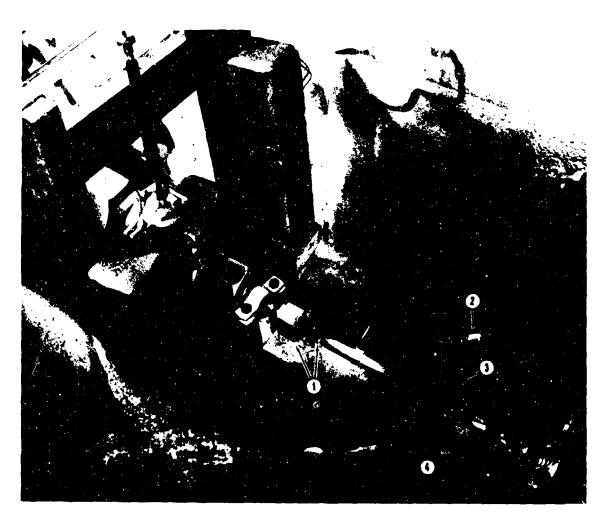
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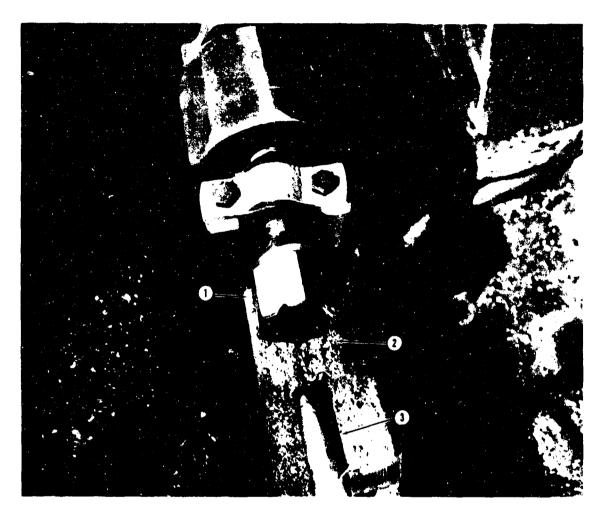
TEST PIPE AFTER EXPLOSION FIGURE 15

- 1) BURNED UNION AND TEST PIPE FITTING
- 2) CHARRED, BENT FLAME PROBE
- 3) BENT THERMOCOUPLE CONNECTOR
- 4) MELTED, EMPTY PRESSURE TRANSDUCER BOSS
- 5) DISPLACED STRAIN GAUGE TERMINALS
- 6) DISPLACED COPPER O-RING ON PRESSURE TRANSDUCER
- 7) RELIEF VALVE



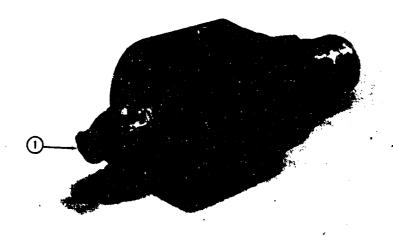
TEST PIPE AFTER EXPLOSION FIGURE 16

- 1) MELTED UNION AND TEST PIPE FITTING
- 2) BROKEN THERMOCOUPLE
- 3) MELTED, EMPTY PRESSURE BOSS
- 4) DISPLACED STRAIN GAUGE TERMINAL



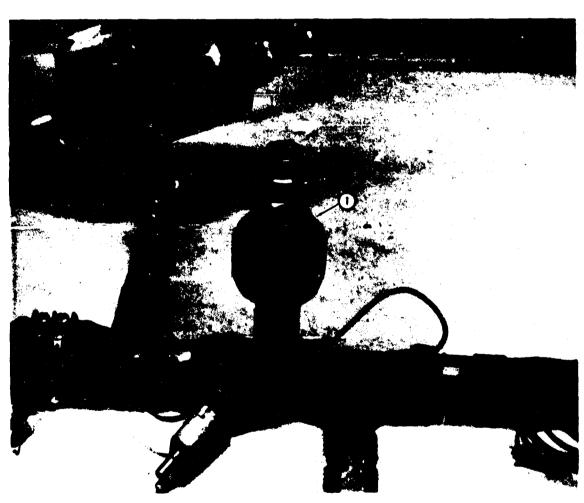
TEST PIPE AFTER EXPLOSION FIGURE 17

- 1) MELTED UNION
- 2) MELTED TEST PIPE FITTING FRAGMENTS
- 3) TEST PIPE



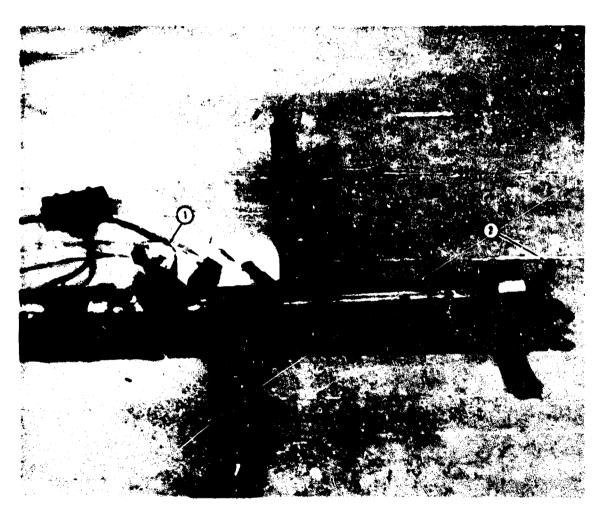
PRESSURE TRANSDUCER AFTER EXPLOSION FIGURE 18

1) PRESSURE TRANSDUCER FITTING



PRESSURE TRANSDUCER AFTER EXPLOSION FIGURE 19

1) DISPLACED, BENT COPPER O-RING ON PRESSURE TRANSDUCER



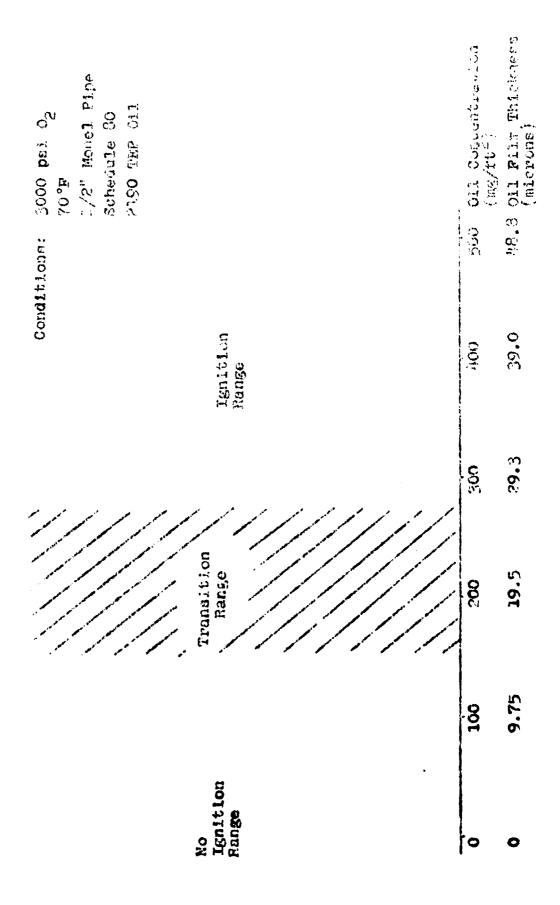
INSTRUMENTATION BOSS AFTER EXPLOSION FIGURE 20

- 1) MELTED PRESSURE TRANSDUCER BOSS
- 2) MELTED TEST PIPE FITTING

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